



Use of an Acetylenic Sulfone as an Alkene Dipole Equivalent in the Synthesis of (±)-Pumiliotoxin C

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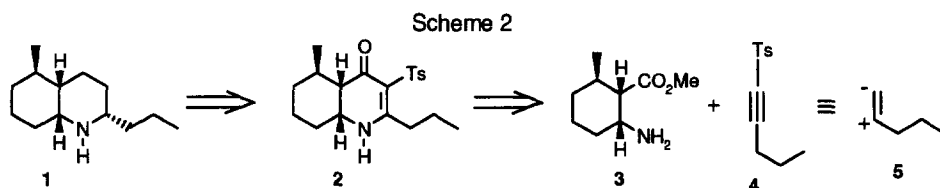
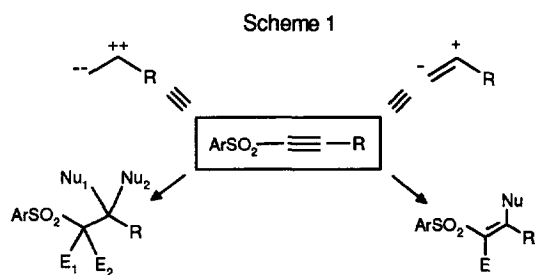
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Abstract: The cycloaddition of methyl *cis*-2-amino-*trans*-6-methylcyclohexanecarboxylate (**3**) with 1-*p*-(toluenesulfonyl)-1-pentyne (**4**) afforded the corresponding enaminone **2**, that was in turn reduced to (±)-pumiliotoxin C (**1**). The acetylenic sulfone **4** thus functions as the synthetic equivalent of the alkene dipole **5** in this process. © 1997, Elsevier Science Ltd. All rights reserved.

Acetylenic sulfones are versatile synthetic reagents¹ that can undergo one or more conjugate additions at the β -position, followed by reaction with one or more electrophiles at the α -position. The sulfone group can be removed reductively at the end of the desired transformations. Thus, acetylenic sulfones act as the

synthetic equivalents of dipoles or "multipoles" (Scheme 1).² When the nucleophile in the initial conjugate addition is an amine, the product is an enamine sulfone³ that in turn has nucleophilic character at the α -position, providing a potential site for alkylation or acylation. The retrosynthesis in Scheme 2 shows how these features can be combined in the novel stepwise cycloaddition of β -amino ester **3** with acetylenic sulfone **4** (Ts = *p*-toluenesulfonyl) to afford enaminone **2**, which then

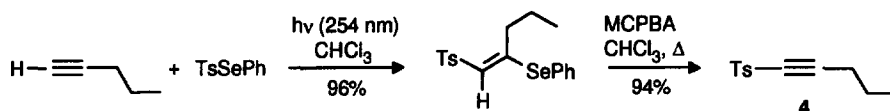
serves as a key intermediate for the preparation of pumiliotoxin C (**1**), an alkaloid that occurs in the skin secretions of the Panamanian poison-dart frog *Dendrobates pumilio*.^{4,5} Thus, **4** behaves as the equivalent of the alkene dipole **5** in the cycloaddition.



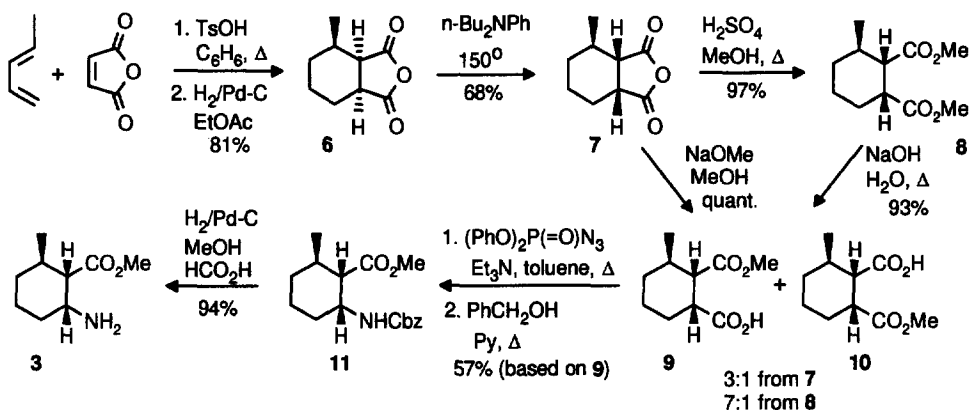
The acetylenic sulfone **4** was prepared by the selenosulfonation of 1-pentyne, followed by selenoxide elimination (Scheme 3), using a procedure reported previously for the selenosulfonation of other acetylenes.⁶

The β -amino ester **3** was obtained as shown in Scheme 4. The cyclic anhydride **7** was prepared by the Diels-Alder cycloaddition of piperylene with maleic anhydride, followed by hydrogenation to **6**, and base-catalyzed epimerization to **7**.⁷ The direct methanolysis of anhydride **7** proceeded with poor regioselectivity to furnish the half esters **9** and **10** in the ratio of 3:1. However, selective partial saponification of the diester **8** afforded **9** and **10** in an improved ratio of 7:1. Since the half-esters **9** and **10** were difficult to separate, the 7:1 mixture was subjected to a Curtius rearrangement effected with diphenylphosphoryl azide,⁸ and the resulting isocyanates were trapped *in situ* with benzyl alcohol to afford the Cbz-protected amine **11**, which was easily separated from the regioisomer derived similarly from **10**. Finally, hydrogenolysis produced the desired amino ester **3**.⁹

Scheme 3

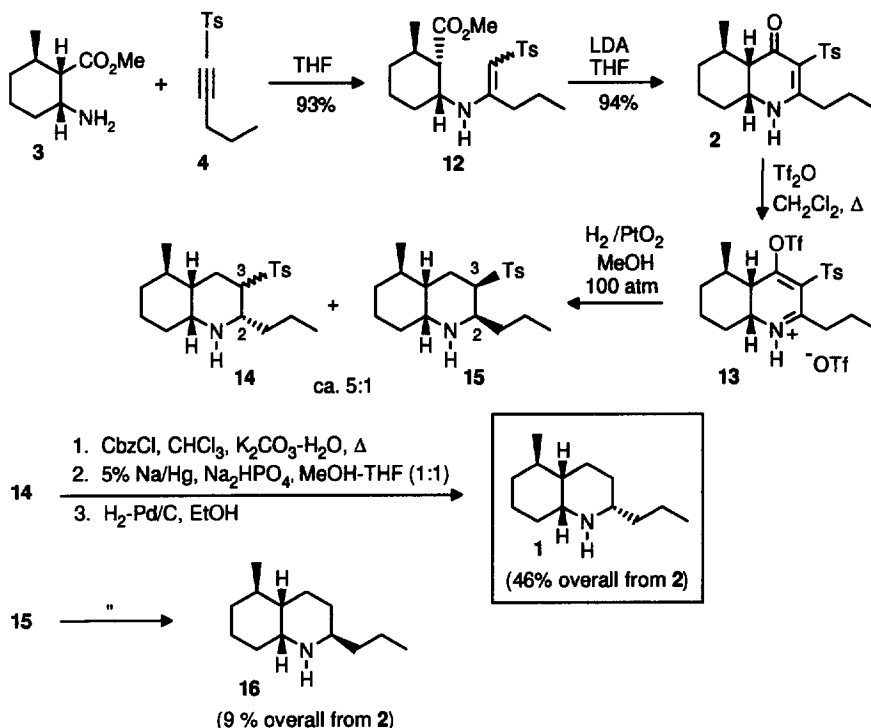


Scheme 4



The conjugate addition of **3** to **4** in THF at room temperature afforded a 2:1 mixture of the *Z* and *E* isomers of **12**, respectively. Ring-closure was effected in 94% yield (based on the recovery of 33% of unreacted starting material) by treatment with LDA to provide the enamionone **2** (Scheme 5).¹⁰ Hydrogenation of the corresponding enol triflate **13** occurred stereoselectively from the *exo* side, producing the C-2 epimers **14** and **15** in the ratio of ca. 5:1. The desired C-2 isomer **14** was obtained as an unseparable mixture of C-3 epimers, whereas the 2-*epi* derivative **15**, which was separated chromatographically from **14**, was a single C-3 stereoisomer, tentatively identified as the 3-*exo* sulfone. Reductive desulfonation¹¹ of the Cbz derivatives of **14** with 5% sodium amalgam, followed by hydrogenolysis of the Cbz group, afforded (\pm)-pumiliotoxin C (**1**) in 46% overall yield from enamionone **2**. Similarly, the N-Cbz derivative of **15** gave (\pm)-2-epipumiliotoxin C (**16**) in 9% overall yield from **2**. Both products **1** and **16** had properties^{12,13} in close agreement with the literature.⁵ The present synthesis of pumiliotoxin C illustrates how an acetylenic sulfone can be employed as an alkene dipole synthon in a novel cyclization of a β -amino ester to a synthetically useful enamionone.

Scheme 5



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References and Notes:

- (a) Simpkins, N.S. *Sulphones in Organic Synthesis*, Pergamon Press: Oxford, 1993. (b) *The Chemistry of Sulphones and Sulfoxides*, Patai, S.; Rappoport, Z.; Stirling, C.J.M., Eds.; Wiley: Chichester, 1988.
- Back, T.G.; Wehrl, D. *Tetrahedron Lett.* **1995**, *36*, 4737.
- For previous examples of conjugate additions of nitrogen nucleophiles to acetylenic sulfones, see: (a) Back, T.G.; Collins, S.; Law, K.-W. *Can. J. Chem.* **1985**, *63*, 2313. (b) Cinquini, M.; Cozzi, F.; Pelosi, M. *J. Chem. Soc., Perkin Trans. 1* **1979**, 1430. (c) Sanders, J.A.; Hovius, K.; Engberts, J.B.F.N. *J. Org. Chem.* **1974**, *39*, 2641. (d) Truce, W.E.; Onken, D.W. *J. Org. Chem.* **1975**, *40*, 3200. (e) Truce, W.E.; Markley, L.D. *J. Org. Chem.* **1970**, *35*, 3275. (f) McDowell, S.T.; Stirling, C.J.M. *J. Chem. Soc. (B)* **1967**, 351. (g) Stirling, C.J.M. *J. Chem. Soc.* **1964**, 5863.
- For isolation and structure determination, see: Daly, J.W.; Tokuyama, T.; Habermehl, G.; Karle, I.L.; Witkop, B. *Liebigs Ann. Chem.* **1969**, 729, 198.
- For previous racemic syntheses, see: (a) Mehta, G.; Praveen, M. *J. Org. Chem.* **1995**, *60*, 279. (b) Meyers, A.I.; Milot, G. *J. Am. Chem. Soc.* **1993**, *115*, 6652. (c) Polniaszek, R.P.; Dillard, L.W. *J. Org. Chem.* **1992**, *57*, 4103. (d) Brandi, A.; Cordero, F.M.; Goti, A.; Guarna, A. *Tetrahedron Lett.* **1992**, *33*, 6697. (e) Comins, D.L.; Dehghani, A. *Tetrahedron Lett.* **1991**, *32*, 5697. (f) LeBel, N.A.; Balasubramanian, N. *J. Am. Chem. Soc.* **1989**, *111*, 3363. (g) Abe, K.; Tsugoshi, T.; Nakamura, N. *Bull. Chem.*

- Soc. Jpn.* **1984**, *57*, 3351. (h) Maruoka, K.; Miyazaki, T.; Ando, M.; Matsumura, Y.; Sakane, S.; Hattori, K.; Yamamoto, H. *J. Am. Chem. Soc.* **1983**, *105*, 2831. (i) Overman, L.E.; Jessup, P.J. *J. Am. Chem. Soc.* **1978**, *100*, 5179. (j) Ibuka, T.; Mori, Y.; Inubushi, Y. *Chem. Pharm. Bull.* **1978**, *26*, 2442. (k) Oppolzer, W.; Fehr, C.; Warneke, J. *Helv. Chim. Acta* **1977**, *60*, 48. (l) Habermehl, G.; Andres, H.; Miyahara, K.; Witkop, B.; Daly, J.W. *Liebigs Ann. Chem.* **1976**, 1577. (m) Oppolzer, W.; Fröstl, W.; Weber, H.P. *Helv. Chim. Acta* **1975**, *58*, 593. (n) Ibuka, T.; Masaki, N.; Saji, I.; Tanaka, K.; Inubushi, Y. *Chem. Pharm. Bull.* **1975**, *23*, 2779. For syntheses of the natural (-)-1, see: (o) Naruse, M.; Aoyagi, S.; Kibayashi, C. *J. Chem. Soc., Perkin Trans. 1* **1996**, 1113. (p) Comins, D.L.; Dehghani, A. *J. Chem. Soc., Chem. Commun.* **1993**, 1838. (q) Murahashi, S.; Sasao, S.; Saito, E.; Naota, T. *Tetrahedron* **1993**, *49*, 8805. (r) Bonin, M.; Royer, J.; Grierson, D.S.; Husson, H.-P. *Tetrahedron Lett.* **1986**, *27*, 1569. (s) Oppolzer, W.; Flaskamp, E. *Helv. Chim. Acta* **1977**, *60*, 204. For syntheses of the unnatural (+)-1, see: (t) Toyota, M.; Asoh, T.; Fukumoto, K. *Tetrahedron Lett.* **1996**, *37*, 4401. (u) Schultz, A.G.; McCloskey, P.J.; Court, J.J. *J. Am. Chem. Soc.* **1987**, *109*, 6493. (v) Masamune, S.; Reed, L.A., III; Davis, J.T.; Choy, W. *J. Org. Chem.* **1983**, *48*, 4441.
6. (a) Back, T.G.; Collins, S.; Kerr, R.G. *J. Org. Chem.* **1983**, *48*, 3077. (b) Back, T.G.; Collins, S.; Gokhale, U.; Law, K.-W. *J. Org. Chem.* **1983**, *48*, 4776. (c) Miura, T.; Kobayashi, M. *J. Chem. Soc., Chem. Commun.* **1982**, 438.
 7. Craig, D. *J. Am. Chem. Soc.* **1950**, *72*, 1678.
 8. Shioiri, T.; Ninomiya, K.; Yamada, S. *J. Am. Chem. Soc.* **1972**, *94*, 6203.
 9. For a different synthesis of **3**, see Davies, S.G.; Bhalay, G. *Tetrahedron: Asymmetry* **1996**, *7*, 1595.
 10. New compounds **2** and **12** were fully characterized by their IR, NMR and mass spectra, and gave satisfactory elemental analyses. The *Z*-isomer of **12** was identified by an NOE between the olefinic CH and allylic CH₂ groups.
 11. For examples of reductive desulfonylations with Na/Hg, see: Trost, B.M.; *Bull. Chem. Soc. Jpn.* **1988**, *61*, 107, and references cited therein.
 12. (±)-Pumiliotoxin C (**1**): oil; IR (film) 3300, 1117, 1089, 754 cm⁻¹; ¹H-NMR (400 MHz) δ 2.87-2.83 (m, 1 H), 2.58-2.50 (m, 1 H), 2.00-1.80 (m, 2 H), 1.75-0.88 (m, 15 H), 0.91 (t, J= 6.9 Hz, 3 H), 0.85 (d, J= 6.6 Hz, 3 H); ¹³C-NMR (100 MHz) δ 57.8, 56.0, 42.6, 39.6, 35.9, 33.3, 27.4, 27.2, 27.0, 21.2, 19.9, 19.1, 14.3; mass spectrum, *m/z* (relative intensity, %) 195 (M⁺, 4), 194 (8), 166 (60), 152 (100). (±)-(**1**) Hydrochloride: mp 242-245 °C (sealed capillary), lit.^{5m} mp 245 °C, lit.⁵ⁿ mp 232 °C; ¹H-NMR (400 MHz) δ 9.60 (m, 1 H), 8.30 (m, 1 H), 3.33 (m, 1 H), 3.00 (m, 1 H), 2.55-2.30 (m, 2 H), 2.23-1.95 (m, 4 H), 1.92-1.84 (broad d, 1 H), 1.83-1.75 (broad d, 1 H), 1.72-1.35 (m, 6 H), 1.32-1.20 (m, 1 H), 1.04-0.95 (m, 1 H), 0.92 (t, J= 7.4 Hz, 3 H), 0.90 (d, J= 6.2 Hz, 3 H); ¹³C-NMR (100 MHz) δ 60.1, 58.0, 40.9, 34.8, 34.3, 29.1, 27.2, 25.2, 23.1, 20.6, 19.7, 19.1, 13.7.
 13. (±)-2-Epipumiliotoxin C (**16**): oil; IR (film) 3352, 1145, 1080, 742 cm⁻¹; ¹H-NMR (400 MHz) δ 3.13 (dt, J= 10.5, 4.1 Hz, 1 H), 2.84-2.75 (m, 1 H), 1.90-1.01 (m, 17 H), 0.99 (d, J= 7.2 Hz, 3 H), 0.91 (t, J= 7.0 Hz, 3 H); ¹³C-NMR (100 MHz) δ 50.0, 49.5, 42.0, 38.4 (br), 32.5 (br), 31.6 (br), 29.7 (br), 28.4 (br), 25.3 (br), 20.5, 19.4, 19.3, 14.2; mass spectrum, *m/z* (relative intensity, %) 195 (M⁺, 2), 194 (3), 166 (5), 152 (100).

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